

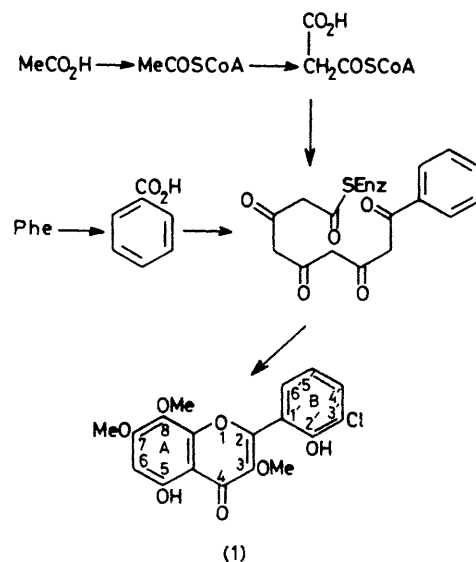
Biosynthesis of Chlorflavonin in *Aspergillus candidus*: A Novel Fungal Route to Flavonoids

By MICHAEL K. BURNS, JANE M. COFFIN, ITSUO KUROBANE, and LEO C. VINING
(Biology Department, Dalhousie University, Halifax, Nova Scotia, B3H 4J1, Canada)

Summary The distribution of radioactivity in chlorflavonin from cultures of *Aspergillus candidus* supplemented with [1-¹⁴C]acetate, [α-¹⁴C]cinnamate, and [7-¹⁴C]benzoate indicates that this flavonoid antibiotic is biosynthesized from one C₆C₁ and four C₂ precursors.

AMONG the large class of flavonoid natural products chlorflavonin (1) and the congeneric dechloro-compound are unique in their fungal origin.^{1,2} As with the flavonoids of higher plants,³ phenylalanine is a biosynthetic precursor and chemical degradation of the product labelled from L-[β-¹⁴C]-phenylalanine located all of the radioactivity at C-2.² [1-¹⁴C]Acetate, a specific precursor of ring A of the plant flavonoids, is incorporated less efficiently than L-[β-¹⁴C]-phenylalanine into chlorflavonin (specific incorporation 2.5 vs. 9.9%). Cinnamate is a known intermediate on the route from phenylalanine to the plant flavonoids. Radioactivity from [α-¹⁴C]- and [β-¹⁴C]-cinnamate was incorporated into chlorflavonin with an efficiency similar to that of [1-¹⁴C]acetate.

To ascertain whether *A. candidus* possesses a novel biosynthetic pathway, samples of chlorflavonin derived from [1-¹⁴C]acetate, [α-¹⁴C]cinnamate, and [β-¹⁴C]cinnamate were degraded with alkali² and the distribution of radioactivity into 4,5-dimethoxyresorcinol and 3-chlorosalicylic acid, derived from rings A and B, respectively, was measured. The Table shows that the label from [1-¹⁴C]-acetate was not present exclusively in ring A and that [α-¹⁴C]cinnamate labelled ring A heavily. The results can be accounted for if C-3 and C-4, as well as ring A originate



SCHEME. Proposed route for the biosynthesis of chlorflavonin.

from acetate. Thus the distribution of the label from [α-¹⁴C]cinnamate is due to its prior metabolism to [¹⁴C]-acetate; the reduced incorporation efficiency of acetate compared with phenylalanine is attributed to stronger metabolic competition for the precursor.

TABLE. Distribution (%) of radioactivity in the degradation products of chlorflavonin.

Precursor	4,5-Dimethoxy- resorcinol	3-Chloro- salicylic acid	Barium carbonate ^a
L- $[\beta\text{-}^{14}\text{C}]$ Phenyl- alanine	Nil	102	102
$[\beta\text{-}^{14}\text{C}]$ Cinnamate	Nil	103	102
$[\alpha\text{-}^{14}\text{C}]$ Cinnamate	48.2	3.7	—
$[7\text{-}^{14}\text{C}]$ Benzoate	0.13	97.6	94.0
$[1\text{-}^{14}\text{C}]$ Acetate	66.5	1.35	—

^a Obtained by decarboxylation of 3-chlorosalicylic acid.

Specific incorporation of $[7\text{-}^{14}\text{C}]$ benzoate into chlorflavonin (Table) is consistent with such a pathway. In a parallel experiment no radioactivity was incorporated from $[7\text{-}^{14}\text{C}]$ salicylic acid, suggesting the route in the Scheme as the most likely for the biosynthesis of flavonoids in *A. candidus*. Here, as in the pathways used to make certain quinonoid secondary metabolites⁴ plants and fungi seem to have discovered independent solutions to a biosynthetic problem.

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⁴ R. H. Thomson, 'Naturally Occurring Quinones,' Academic Press, London, 1971.